Photochemical Alkylation of Caffeine with Amino-acids

By D. ELAD* and I. ROSENTHAL

(Department of Chemistry, The Weizmann Institute of Science, Rehovoth, Israel)

Summary Photochemical reactions of caffeine with aliphatic α -amino-acids lead to 8-alkylcaffeine derivatives.

Photochemical reactions of purines with alcohols have been described recently. These reactions result in addition of the alcohol across the 1,6-double bond or substitution at the 8-position of the purine ring system. We report photochemical reactions of caffeine with various aliphatic α -amino-acids which lead to caffeine derivatives with different alkyl substituents in the 8-position of the

caffeine molecule. These reactions can be summarized as follows:

$$\begin{array}{c|c}
 & Me \\
 & Me \\
 & N \\
 & N$$

The reactions studied and the major products isolated are described in the Table.

TABLE Photochemical alkylation of caffeine with amino-acids

Amino acid			Product 8-substituted caffeine	Yield ^a
Alanine			Et	21%
2-Aminobutyric	acid		$\begin{cases} \text{Et} \\ \text{Pr} \end{cases}$	$\left. egin{array}{c} 5 \ 22 \end{array} ight\}$
Serine			Et	21
Threonine	• •		$\begin{cases} \text{Et} \\ \text{Pr} \end{cases}$	$\{12, 10\}$
Norvaline			Bu	22
Leucine		• •	$Me_2CH[CH_2]_2$	19
Norleucine			Bu-CH_2	22
Alanylglycine	• •		Et	21

^a Yields are based on caffeine used.

Typically, a solution of caffeine (500 mg.) and DL-alanine (500 mg.) in water (70 ml.) was irradiated under nitrogen for 24 hr. The reaction mixture was concentrated to about half of its volume and extracted with chloroform. The organic solvent was evaporated and the residue was chromatographed on silica gel to isolate 8-ethylcaffeine (122 mg.). Products were isolated and identified by elemental analysis, n.m.r. and mass spectra, and in some cases by comparison with authentic samples.3

It has been reported that the interaction of photoactivated adenosine with glycine leads to the fragmentation of the latter and the formation of free radicals derived from the amino-acid.4 A similar mechanism might be operating in the reported reactions, where caffeine serves as the lightabsorbing system and the excited caffeine molecule induces fragmentation of the amino-acid to reactive species.5

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[†] Hanovia 450 w high pressure mercury vapour lamp with Corex filter.

¹ H. Linschitz and J. S. Connolly, J. Amer. Chem. Soc., 1968, 90, 2979; J. S. Connolly and H. Linschitz, Photochem. and Photobiol., 1968, 7, 791.

² D. Elad, I. Rosenthal, and H. Steinmaus, *Chem. Comm.*, 1969, 305.

² E. S. Golovchinskaya, Sbornik Statei obshchei Khim., Akad. Nauk S.S.S.R., 1953, 1, 692 (Chem. Abs., 1955, 49, 1070a). ⁴ R. Santus, C. Helene, and M. Ptak, Compt. rend., 1966, 262, D, 2077.

⁵ Cf. R. Noyori, M. Kato, M. Kawanishi, and H. Nozaki, Tetrahedron, 1969, 25, 1125.